



**University of Vigo and JSPS Core-to  
Core Program Joint Seminar**

**“Atomically  
Controlled Processing  
for Ultralarge Scale  
Integration”**

**Vigo, September 4<sup>th</sup>-6<sup>th</sup>, 2012**



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**General information on research lines of possible interest at the**

**Laser group of the University of Vigo**

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Since the fabrication of the first lasers around 50 years ago, a large variety of improved laser systems delivering laser radiation with wavelength from deep ultra violet (UV) to far infra red (IR) have been developed. Radiation can nowadays be applied to materials continuously or with short pulses of down to few femtoseconds duration and applications are nowadays focused both to R&D and commercial issues, satisfying a wide range of purposes in military, scientific, medical and uncountable industrial fields. The exigencies of the scientific community to exceed detection limits in the analysis of material properties and to deliver photons with extraordinary accuracy in terms of photon energy, spatial coherence and power, together with the industrial demand of having cheap, high power equipments that are reliable and easily to maintain have boosted the improvement of commercial laser systems.

The mission of the “Laser group” at the University of Vigo with its 3 sub-divisions “Industrial applications of Lasers Group”, “Metrology Group” and “New Materials Group” is focused on the development of production and detection processes based on commercial laser systems. This introduction with very general information on the main research lines of the laser group will anticipate extensive lab visits, where infrastructure, basic concepts and recent achievements of different laser assisted processing techniques, such as Laser induced CVD (LCVD), Pulsed Laser Deposition (PLD), Surface functionalization, Excimer Laser Assisted Annealing (ELA) and Crystallization (ELC), Pulsed Laser Induced Epitaxy (PLIE), Laser cleaning, Laser cladding, Laser spinning, Welding, cutting and drilling, as well as TV holography and interferometry will be shown in detail.

Such material processing must inevitably be accompanied by and receive back-up from exhaustive characterization of surfaces and interfaces as well as by fundamental studies of the interactions between laser radiation and the irradiated materials. An overview of the facilities, available in the groups laboratories as well as at the central facilities of the University of Vigo, will be also given introduce the corresponding lab visits with practical demonstrations of the available techniques.

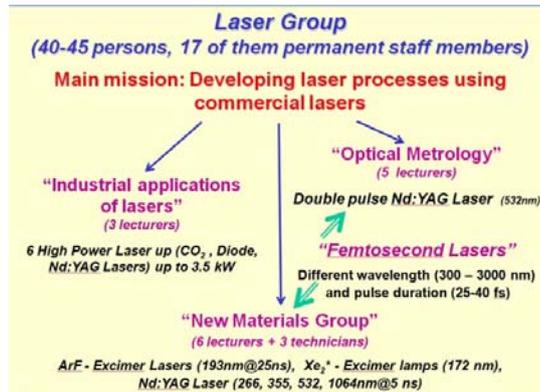


Fig.1 Laser group at UVigo and equipments



Fig.2 Some research topics of the "Industrial applications of lasers" group



Fig.3 Some research topics of the "Metrology" group



Fig.4 Some research topics of the "New Materials" group

## LASER PROCESSING AND SYNTHESIS OF MATERIALS AT MICRO- AND NANOSCALE

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The successful application of laser technologies to process a broad diversity of materials can only be understood after the recent development of a likewise broad range of laser sources. Thus, the selection of the laser wavelength, laser pulse energy and duration is a key factor for the development of new processing techniques. Figure 1 enumerates the laser techniques in relation to laser-material interaction time.

These laser based techniques can be classified in different groups as a function of their objective. Hence, there are techniques whose purpose is the production of a coating onto a substrate, such as the Pulsed Laser Deposition (PLD) and laser cladding. Our innovations in the field of these techniques start with the production of biocompatible coatings such as hydroxyapatite or bioactive glasses onto metal substrates for implants. More recently, the capabilities of laser cladding have been stretched with our development of an unique facility for microcladding. Laser microcladding of metallic alloys onto metals and copper onto crystalline silicon are some examples of the potential of this technique.

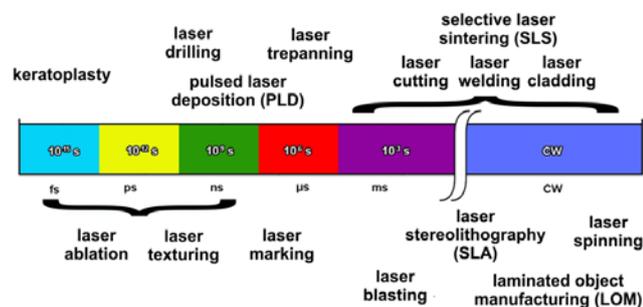


Fig 1. Laser assisted processes as a function of pulse duration.

On the other hand, we have available laser based techniques related to surface modification of materials, such as laser marking, laser texturing or laser blasting. The main purpose of laser texturing and laser blasting is to change the surface roughness and wettability of materials we stretched. Although recent research covers biocompatible polymers (see figure 2), these techniques have been applied mostly on metallic implants such as titanium and titanium alloys.

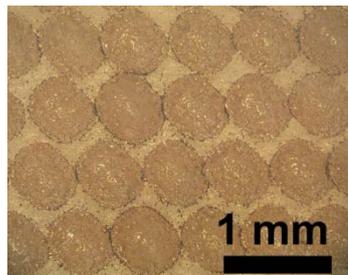
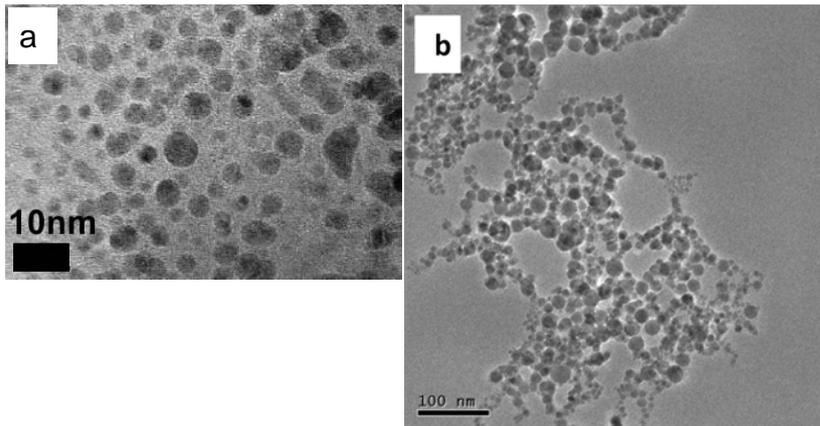


Fig. 2. Laser texturing of PEEK.

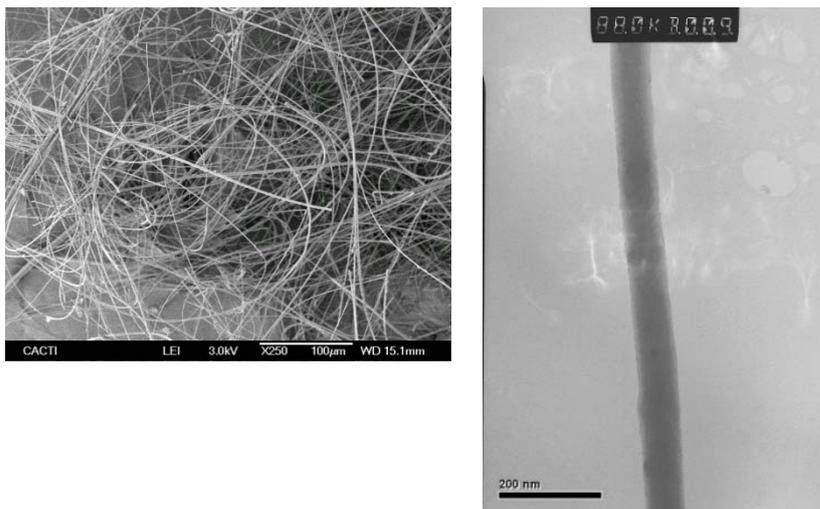
Techniques aimed to produce three dimensional structures are classified into additive manufacturing procedures. Here we can include all laser based rapid prototyping techniques such as Selective Laser Sintering (SLS), Laser Stereolithography (SLA), and Laminated Object Manufacturing (LOM).

Referring to the capability of lasers to synthesize nanomaterials, we can highlight two techniques: laser ablation for production of nanoparticles and laser spinning that has been recently introduced for the production of glass nanofibers.  $\text{TiO}_2$  nanoparticles with controllable average diameter have been obtained by Continuous Wave (CW) laser ablation of metallic Ti submerged in water. The use of a CW laser contributes to a complete reaction between the metallic species and the evaporated liquid due to long interaction time. Obtained nanoparticles are almost perfect spheres in shape with a stoichiometric composition corresponding to  $\text{TiO}_2$ . Furthermore, this technique has been probed to produce hydroxyapatite nanoparticles by laser ablation in ambient conditions. Figure 3.a shows a sample of these hydroxyapatite nanoparticles, while figure 3.b shows the titanium oxide nanoparticles.



**Fig. 3.** TEM micrographs showing a) HA and b) TiO<sub>2</sub> nanoparticles produced by laser ablation.

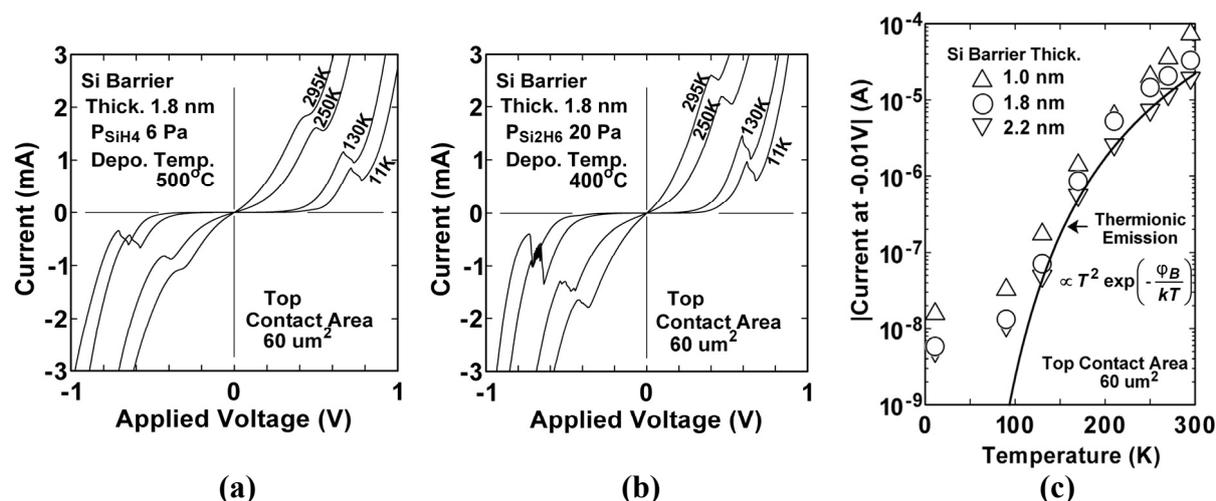
Laser spinning is a new technique which allows the production of ultralong glass nanofibers. This technique opens up new possibilities to produce glass nanofibers that have not been obtained with other methods. Figure 4 shows 45S5 Bioglass® nanofibers produced by Laser spinning. This bioactive glass nanofibers can be employed to build scaffolds for tissue engineering. Additionally, their application for bone defect filling is advantageous compared to granules, thanks to their high length, flexibility and mechanical resistance which make possible to adapt them to any form and insert or remove easily. Also, nanofibers of different functional compositions such as aluminosilicates for refractory applications or lithium silicates for carbon capture have been produced.



**Fig. 4.** SEM and TEM micrographs showing and overall view and a detail of the micro- and nanofibers of 45S5 Bioglass® produced by Laser spinning.

In summary, lasers are available today to treat, process, or synthesise many micro- and nanomaterials, opening a broad field of possibilities worth to be explored and applied.





**Fig. 2.** (a) Current-voltage characteristics of RTDs with 1.8 nm-thick Si barriers for various measurement temperatures. Si barrier growth was performed by reaction of (a)  $\text{SiH}_4$  at 500 °C and (b)  $\text{Si}_2\text{H}_6$  at 400 °C. (c) Temperature dependence of non-resonance current (at -0.01V) for the RTDs with 1.0, 1.8 and 2.2 nm-thick Si barriers. Calculated fitting values based on thermionic-emission are also shown by a solid line.  $\phi_B$ ,  $k$  and  $T$  in the equation are effective barrier height, Boltzmann constant and temperature, respectively.

As shown in current-voltage characteristics of RTDs (**Fig. 2 (a),(b)**), it has been clarified that lowering the Si barrier growth temperature down to 400 °C enables to achieve improved NDC characteristics at around room temperature. Temperature dependence of the non-resonance current shows existence of non-thermal leakage current (observed at lower temperatures below 100 K) and the current tends to increase with decrease of Si barrier thickness. It is concluded that improvement in heterointerface flatness is necessary for the NDC enhancement in nanometer-order thin Si barrier RTD. Additionally, thermionic-emission dominant region (observed at higher temperatures above 100 K in **Fig. 2 (c)**) indicates a possibility that introduction of larger barrier height (i.e. larger band discontinuity) enhances the NDC at room temperature by suppression of thermionic-emission current. The thickness dependence indicates that thermionic-emission current is sensitive to the nanometer-order Si barrier thickness. Therefore, it is confirmed that suppression of roughness generation is indispensable for high reproducibility of the resonant tunneling diodes.

From these results, it is found that only about 1 nm thick Si layer acts as barrier for RTD and there is a possibility that heavy atomic-layer doping of impurity (e.g. C, N and so on) might strongly influence barrier properties for resonant tunneling. Therefore, low-temperature epitaxial growth process (e.g. extremely low-temperature thermal CVD and low-energy plasma CVD [6]) becomes increasingly important to modulate electronic properties of nanometer-order ultrathin layers of group IV semiconductor far from thermal equilibrium.

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## Charge Storage and Optoelectronic Response of Silicide-Nanodots/Si-Quantum-Dots Hybrid-Floating-Gate MOS Devices

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The application of silicon quantum dots (Si-QDs) and metallic nanodots (NDs) to a floating gate (FG) in MOS devices has attracted much attention because of their potential advantages over conventional FG MOS memories. In fact, multi-valued memory capability can be provided by discrete charged states of Si-QDs originating from the quantum confinement energy as demonstrated in a unique stepwise shift in the threshold voltage of MOSFETs with a Si-QDs floating gate [1-4]. On the other hand, enlarged and stabilized memory window can be achieved by a FG made of metallic NDs with an appropriate work function to form a deep potential well [5, 6]. Recently, we have proposed and fabricated hybrid stacked structures consisting of metallic NDs, Si-QDs and ultrathin interlayer SiO<sub>2</sub> to satisfy both multiple valued capability and charge storage capacity for a sufficient memory window and to open up novel functionality [7]. As for the functionality of such a hybrid nanodots FG, with near-infrared light irradiation, we have verified optical responses caused by transfer of photo-excited electrons from metallic NDs to Si-QDs [8].

Hybrid stacked structures consisting of NiSi-NDs, ultrathin interlayer SiO<sub>2</sub> and Si-QDs were fabricated through the following process sequence. Hemispherical Si-QDs were firstly formed on an ultra-thin thermally-grown SiO<sub>2</sub> by controlling the early stages of LPCVD of pure SiH<sub>4</sub> at 580°C. The areal dot density and the average dot size evaluated by AFM measurements were typically ~5nm and ~3.5x10<sup>11</sup>cm<sup>-2</sup>, respectively. And then, the Si-QDs surface was slightly oxidized in O<sub>2</sub> at 850°C, and followed by SiO<sub>2</sub> deposition from inductively-coupled remote plasma (ICRP) of SiH<sub>4</sub> and excited O<sub>2</sub>/Ar at 350°C to obtain the designed thickness. Subsequently, to form NiSi-NDs, Si-QDs were grown again under the same conditions as the first formation of Si-QDs, and the surface was covered uniformly with a ~1.8nm-thick Ni layer by electron beam evaporation and successively exposed to remote H<sub>2</sub> plasma without external heating to enhance surface migration of Ni atoms and full-silicidation of Si-QDs. The 3rd formation of Si-QDs was preformed after ICRP-CVD of ultrathin SiO<sub>2</sub> on NiSi-NDs. Lastly, the top control oxide with a thickness of ~20nm, Al gates with a diameter of 1 mm and Al back contact with a window for light irradiation were sequentially fabricated to complete hybrid FG MOS structures (Fig. 1).

High-frequency capacitance-voltage (C-V) characteristics of the MOS capacitor with the hybrid FG show positive and negative flat-band voltage shifts depending on the polarity and the maximum magnitude of applied gate bias (Fig. 2), and confirm stable charge storage in a deep potential well in each of NiSi-NDs. Considering that electrons in NiSi-NDs can be

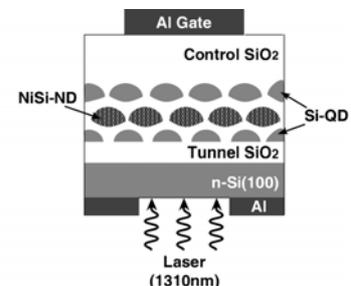


Fig. 1 Schematic of a MOS capacitor with a NiSi-NDs/Si-QDs hybrid FG.

excited by irradiation of infrared light of being not absorbed in the Si substrate and Si-QDs, such an infrared light irradiation can cause electron transfer from NiSi-NDs to Si-QDs selectively (Fig. 3) and induces flat-band voltage shift as a result of the change in the charge centre in the hybride FG. To investigate the response of photoexcited electrons in the hybrid FG stack to the change of gate bias, the transient current induced by pulsed gate voltages was evaluated by connecting the MOS capacitor with a 1 kΩ resistance in series and measuring the voltage drop across it with a lock-in amplifier. As shown in Fig. 4, a distinct change in the output voltage ( $V_{OUT}$ ) of the lock-in amplifier was detected in synchronized timing of the infrared light irradiation under the application of periodic pulsed gate voltage. This result can be attributed to an increase in the displacement current mainly due to the photoexcited electron transfer from the NiSi-NDs to the top and bottom Si-QDs in response to pulsed gate voltage. Namely, the difference in  $V_{OUT}$  ( $\Delta V_{OUT}$ ) between in the dark and under light irradiation corresponds to the amount of charge transferred from the NiSi-NDs to the Si-QDs. The  $\Delta V_{OUT}$  measured in each bias polarity as a function of pulse voltage is summarized in Fig. 5. The  $\Delta V_{FB}$  almost linearly increased with pulse voltage over  $\pm 2.0V$ . This result indicates that the amount of transferred charge in each cycle is limited to a certain level determined by gate voltage and the signals proportional to gate voltage are associated with charging and discharging a capacitor as long as the photoexcited electron transfer can respond to pulsed gate voltage.

**Acknowledgments**

This work was supported in part by Grants-in Aid for Scientific Research (A) No. 24246054 from the Ministry of Education, Culture, Sports, Science and Technology, Japan. In addition, the author deeply appreciates that MOSFETs with hybrid NDs FG were fabricated successfully by utilizing the clean room facilities of Research Institute for Nanodevice and Bio Systems (RNBS), Hiroshima Univ.

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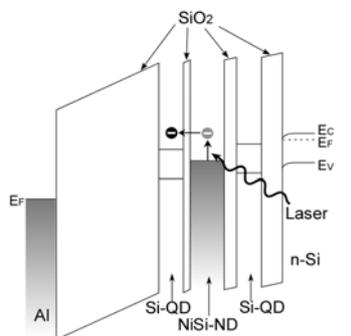


Fig. 3 Energy band diagram of the MOS capacitor with the NiSi-NDs/Si-QDs hybrid floating gate stack in which photoexcited electrons are transferred from the NiSi-NDs to the top Si-QDs with infrared light irradiation.

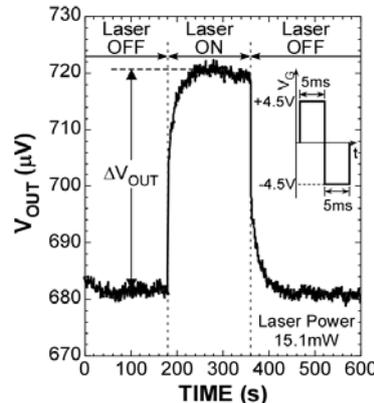


Fig. 4 The response of electrons excited by 1310 nm light irradiation in the NiSi-NDs/Si-QDs hybrid FG to pulsed gate voltage of  $\pm 4.5 V$  at a frequency of 100 Hz.

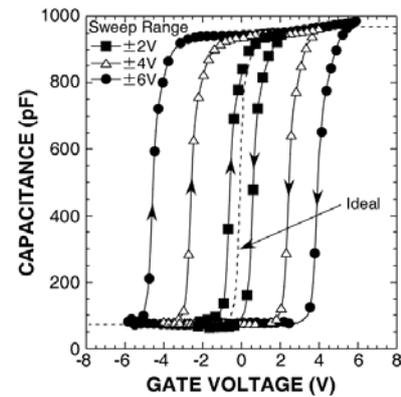


Fig. 2 100 kHz capacitance-voltage (C-V) characteristics of a MOS capacitor with the NiSi-NDs/Si-QDs hybrid FG. The gate voltage sweep rate was set at 100 mV/s. The ideal C-V curve is also shown with a dashed line.

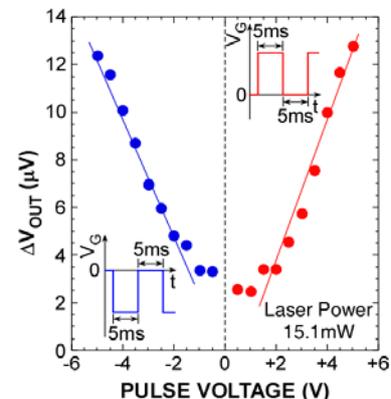


Fig. 5 The response of electrons excited by 1310 nm light irradiation in the NiSi-NDs/Si-QDs hybrid FG to pulsed gate voltage at a frequency of 100 Hz as a function of pulse voltage.

## Optical studies of Ge/Si based heterostructures and Si interstitials related defects in SiGe alloys

J. P. Leitão

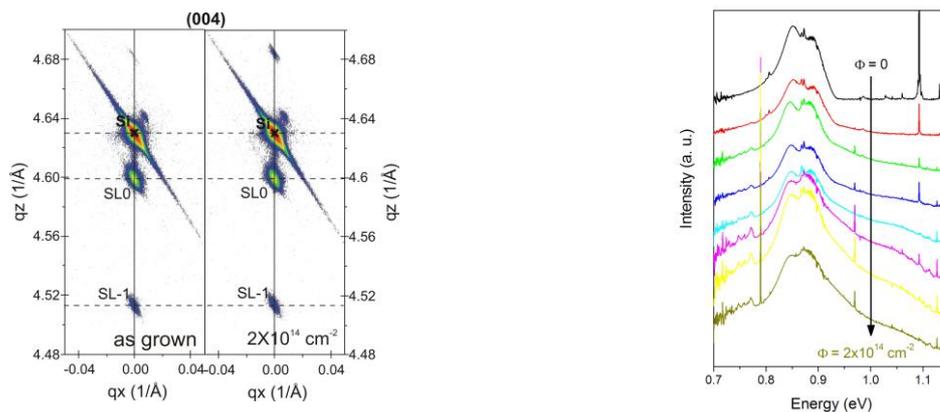
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This work is focused on the application of photoluminescence (PL) for the study of a low dimensional structure based on the Si/Ge system and of Si interstitial related optical centers in Si and SiGe alloys. In the first part, we consider a superlattice structure with embedded Ge islands grown by molecular beam epitaxy (MBE) at 600°C on top of a Si(001) substrate [1]. The samples were irradiated with 2.0 MeV protons to fluences in the range  $2 \times 10^{12}$  -  $2 \times 10^{14}$  cm<sup>-2</sup>. The structural and optical properties of the samples were investigated by cross-sectional transmission electron microscopy, X-ray reciprocal space mapping (RSM), X-ray reflection (XRR) and Rutherford backscattering/channeling. No changes to the as-grown heterostructure were observed after the irradiation in all fluences (Fig. 1). The nominal period of the superlattice was confirmed. The radiative recombination related to type-II transitions in the Si/Ge interface was observed in PL measurements additional to free and bound excitons recombination in the Si layers and substrate, near the energy band gap of Si. The PL related to the Ge islands was observed even for the highest irradiation fluence showing an extremely high radiation hardness of the studied structure, which was confirmed by the absence of changes in the structural properties (Fig. 1).

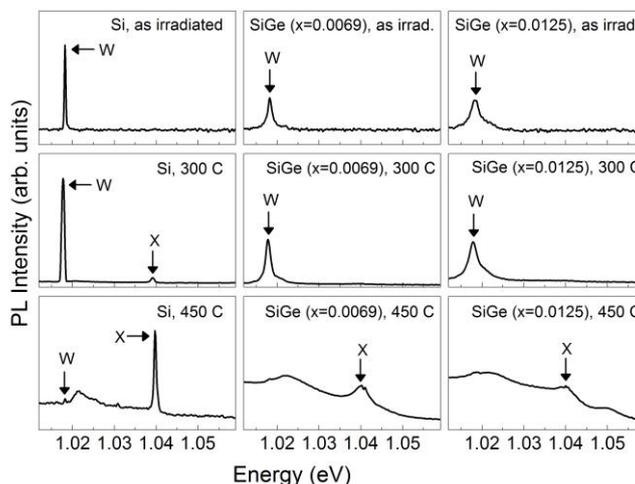
In the second part of the work, the Si interstitial related W and X centers, were studied by photoluminescence and the results compared with the ones from first-principles calculations [2]. The samples were obtained after the growth of a Si or Si<sub>1-x</sub>Ge<sub>x</sub> alloy layers (x=0.0069, 0.0125) on top of a Si(001) substrate at 800°C by MBE. For the production of the centers, irradiation with 875 keV protons (dose of  $10^{16}$  H<sup>+</sup> cm<sup>-2</sup>) followed by annealing of 15 min, in N<sub>2</sub> atmosphere, at a temperature in the range 100 – 650°C, was performed. It was found that the annealing temperature (300°C) that maximizes the PL intensity of the W center is independent of the Ge content, whereas for the X center, a shift was observed from the Si layer (400°C) to the SiGe alloys (450°C) suggesting that the minority Ge atoms delay the formation of this center. The radiative excited state in both centers comprises a pseudodonor state, where a trapped exciton combines a tightly bound hole and a diffuse electron. No acceptor level in the gap was found for either of the centers. The ionization energies of both centers were calculated from a temperature dependence of the PL intensity and the hole binding energy was extracted considering also the spectroscopic energy of the W and X lines in the spectra. The calculated change rates of donor levels with Ge content are in qualitative agreement with the hole binding energy shifts obtained from the experiments. Our results support the previous assignment of the trigonal (I<sub>3</sub>) and tetragonal (I<sub>4</sub>) forms of tri- and tetra-interstitial defects to the W and X centers, respectively.

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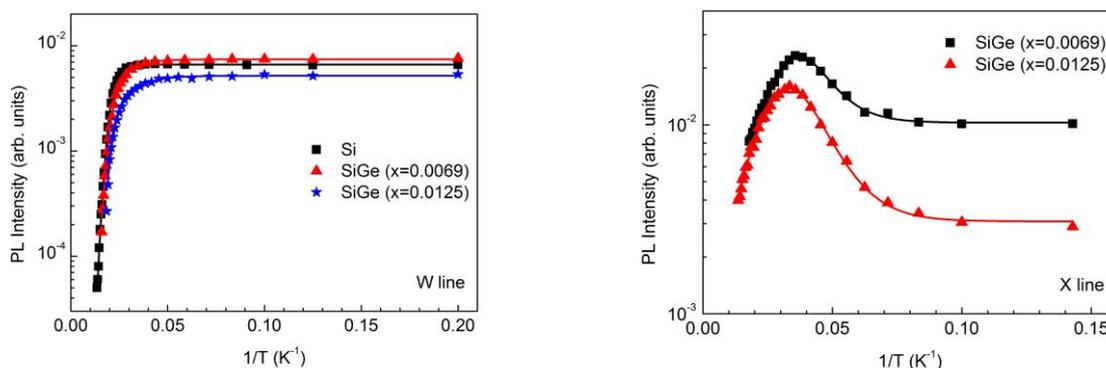
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**Fig. 1** a) X-ray reciprocal space map for the as-grown and irradiated to the highest fluence samples. The Si substrate and superlattice related peaks are identified. b) Photoluminescence spectra of as-grown and irradiated samples (range from  $2 \times 10^{12}$  to  $2 \times 10^{14} \text{ cm}^{-2}$ ).



**Fig. 2.** PL spectra recorded at 5 K for the Si and SiGe layers as irradiated (top row), after an annealing at 300°C (middle row), and after annealing at 450°C (bottom row). Both W and X lines are identified in the spectra.



**Fig. 3** Integrated PL intensity temperature dependence of the W and X lines in Si (squares) and SiGe alloys with  $x=0.0069$ ,  $0.0125$  (triangles and stars, respectively). The lines represent the best fit to the experimental points according to the model described in Ref. [2].

## Low threading dislocation density Ge growth and heavy phosphorus doping in Ge

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Heteroepitaxial growth of Ge on Si has generated great interest for different applications such as optoelectronic devices [1] and novel CMOS technologies [2]. Low threading dislocation density (TDD) is required to prevent degradation of electrical properties of devices, such as dark current of optical detectors. Deposition of Ge on virtual substrate is widely investigated using different approaches to reduce TDD [3-5]. Additionally, N doping such as phosphorus in Ge have been widely investigated because of potential to change Ge to direct band gap material by filling electron into L-valley. In this report, we demonstrate low threading dislocation density Ge growth by combination of cyclic annealing and etching. The heavy P doping in Ge by atomic-layer doping (P-ALD) approach is also discussed.

Epitaxial growth of Ge is carried out using a single wafer reduced pressure CVD system. After HF last clean, the wafer is baked at 850-1000°C and cooled down to 600°C in H<sub>2</sub> and further cooled down to 300°C in N<sub>2</sub> to form hydrogen-free Si surface. Then thin 2-dimensional Ge seed layer (~50 nm) is deposited at 300°C using GeH<sub>4</sub> in N<sub>2</sub> carrier gas. After that the 2nd Ge layer is deposited at 550°C with H<sub>2</sub>-GeH<sub>4</sub> gas mixture. For TDD reduction, cyclic annealing [5] in H<sub>2</sub> is performed during the Ge growth. After the deposition, the Ge layers of some samples are etched back by HCl in the CVD reactor for thinning. For the P-ALD experiment, the Ge surface is exposed to PH<sub>3</sub> with N<sub>2</sub> carrier gas. The exposure temperature is varied between 100-300°C. In order to investigate the effect of hydrogen-termination of the Ge surface, some wafers are cooled down in H<sub>2</sub> environment after the Ge layer deposition. After the exposure step the temperature is changed back to 300°C in N<sub>2</sub> and the Ge layer growth is continued with N<sub>2</sub>-GeH<sub>4</sub> gas mixture. P dose and profile are measured by SIMS. Active P dose is determined by Hall measurement at 10 K.

TDD as a function of blanket Ge thickness is shown in Fig. 1. In the case of the Ge growth with cyclic annealing (○), TDD is decreasing with increasing Ge thickness and  $\sim 7 \times 10^5 \text{ cm}^{-2}$  is achieved for 4.7 μm thick Ge [5]. By 4.5 μm thick Ge deposition and etching process (□), the level of the TDD is lower compared to the TDD of Ge layer of same thickness deposited by cyclic annealing process. For example TDD is reduced to  $\sim 1.3 \times 10^6 \text{ cm}^{-2}$  for a sample which was thinned to 1.8 μm, which is around one order of magnitude lower compared to deposited samples without etching. If the layer is thinned further the TDD is increasing, but is still lower compared to TDD measured for the Ge layers obtained by deposition only.

AFM images of blanket Ge surface before (4.5 μm) and after etching (1.8 μm) are shown in Fig. 2. By etching, cross hatch pattern becomes slightly smeared, but the surface roughness is not increased. Further etching was increasing the surface roughness.

In order to discuss impact of hydrogen-termination on Ge surface, P dose as function of PH<sub>3</sub> exposure temperature is shown in Fig. 3. During the cooling, N<sub>2</sub> or H<sub>2</sub> is chosen as carrier gas. In the case of the sample cooled down in H<sub>2</sub>, very few P is adsorbed on the Ge surface at 100°C. With increasing PH<sub>3</sub> exposure temperature adsorbed P dose is increased. On the other hand, in the case of the sample cooled down in N<sub>2</sub>,  $\sim 8 \times 10^{13} \text{ cm}^{-2}$  of P is adsorbed even at 100°C. In this case, temperature dependence of P dose is small. At 300°C, no clear difference between N<sub>2</sub> and H<sub>2</sub> cooling is observed. These results indicate that the P adsorption is suppressed by hydrogen-termination of the Ge surface in the case of cooling down in H<sub>2</sub>. The hydrogen desorption from the Ge surface starts from  $\sim 200^\circ\text{C}$  resolving the surface passivation [6].

In Fig. 4, P dose as function of PH<sub>3</sub> exposure time at various PH<sub>3</sub> partial pressures is shown.

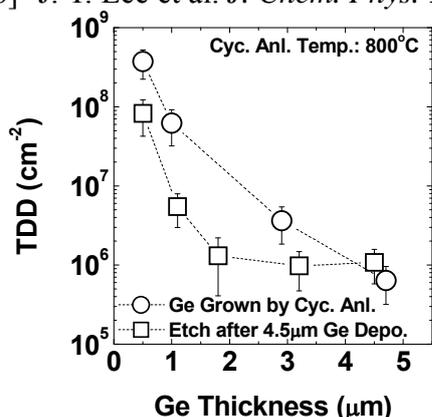
The PH<sub>3</sub> exposure is performed at 300°C on hydrogen-free Ge surface. For all PH<sub>3</sub> partial pressure used, P dose is increasing with increasing PH<sub>3</sub> exposure time and saturated at  $\sim 1.5 \times 10^{14}$  cm<sup>-2</sup>, which value is close to a quarter of monolayer of Ge (100). The saturation value does not depend on PH<sub>3</sub> partial pressure. The incorporated P dose  $Q_r$  on the hydrogen-free Ge surface can be described by Langmuir type kinetic.

$$Q_r = N_t \left[ 1 - \exp \left\{ -k_r \left( \frac{K P_{PH_3}}{1 + K P_{PH_3}} \right) \right\} \right] \quad (1)$$

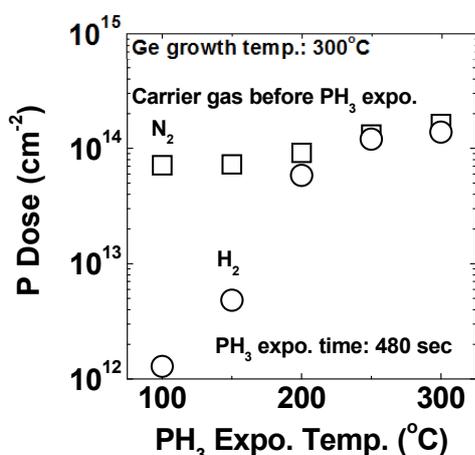
Where  $N_t$  is saturation value of incorporated P dose,  $k_r$  and  $K$  are reaction rate and thermal equilibrium constants of PH<sub>3</sub> adsorbed on the Ge surface, respectively. Good agreement to Langmuir type adsorption model is obtained with  $N_t = 1.55 \times 10^{14}$  cm<sup>-2</sup> (a quarter of monolayer),  $k_r = 77$  s<sup>-1</sup> and  $K = 3.0 \times 10^{-2}$  Pa<sup>-1</sup>.

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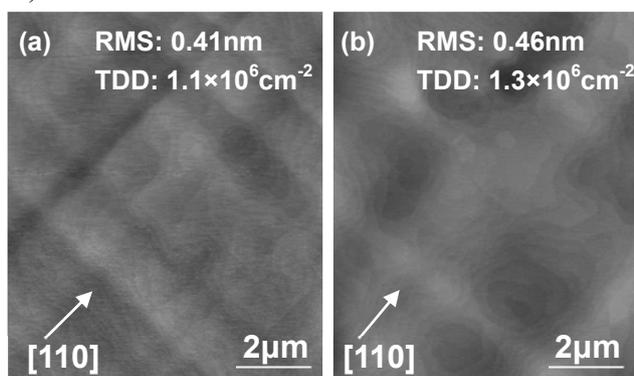
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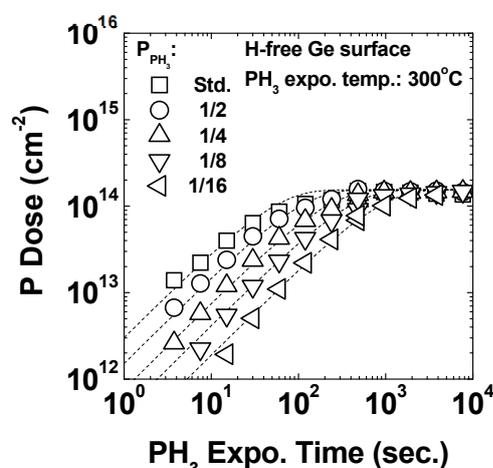
**Fig. 1.** TDD as function of Ge thickness. Blanket Ge is deposited. (○) shows the Ge sample deposited by cyclic annealing process and (□) shows the sample with Ge deposition (4.5 μm) and etching.



**Fig. 3.** P dose as function of PH<sub>3</sub> exposure temperature. N<sub>2</sub> (□) or H<sub>2</sub> (○) is used during temperature change before PH<sub>3</sub> exposure. PH<sub>3</sub> exposure time is 480 sec.



**Fig. 2.** AFM images of the sample after blanket Ge growth with cyclic annealing (a) and followed by etching (b). Ge thicknesses are (a) 4.5 μm and (b) 1.8 μm, respectively. [110] directions are shown by arrows.



**Fig. 4.** P dose as function of PH<sub>3</sub> exposure time. PH<sub>3</sub> partial pressure is varied. Dash lines are calculated by using equation (1) with  $N_t = 1.55 \times 10^{14}$  cm<sup>-2</sup>,  $k_r = 77$  s<sup>-1</sup> and  $K = 3.0 \times 10^{-2}$  Pa<sup>-1</sup>.

## Epitaxial Growth and Characterizations of $\text{Ge}_{1-x}\text{Sn}_x$ and $\text{Ge}_{1-x-y}\text{Si}_x\text{Sn}_y$ Thin Layers for Nanoelectronic and Optoelectronic Applications

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$\text{Ge}_{1-x}\text{Sn}_x$  and  $\text{Ge}_{1-x-y}\text{Si}_x\text{Sn}_y$  alloy thin films are attractive for future Si nanoelectronic and optoelectronic applications such as high carrier mobility MOSFET, photo detector, quantum well laser, photovoltaic cell, and so on. We can engineer the energy band diagram and carrier properties with controlling the lattice constant and strain structures of these thin layers. However, the epitaxial growth of group IV-materials including Sn with its content higher than a few % is not so easy, because Ge-Sn and Si-Sn systems are eutectic, and Sn precipitation easily occurs due to the low equilibrium solid solubility of Sn in Ge and Si. We have developed the low temperature growth technique of  $\text{Ge}_{1-x}\text{Sn}_x$  and the lattice mismatch engineering between  $\text{Ge}_{1-x}\text{Sn}_x$  and various substrates [1-3]. Recently, we achieved the growth of  $\text{Ge}_{1-x}\text{Sn}_x$  epitaxial layers with a Sn content as high as 13% and 27% on Ge and InP substrates, respectively [3].

Engineering the electronic and optical properties of  $\text{Ge}_{1-x}\text{Sn}_x$  is also essentially important for its applications. Figure 1 shows the Hall concentration in undoped  $\text{Ge}_{1-x}\text{Sn}_x$  thin layers grown on Si-on-insulator (SOI) substrates as a function of the Sn content for as-grown,  $\text{N}_2$ -annealed, and  $\text{H}_2$ -annealed samples [4]. We can generally observe p-type conduction even for an undoped Ge epitaxial layer due to unintentional hole generation from vacancy defects in Ge. We found that the Sn incorporation with the content of 0.1% effectively reduces the Hall carrier concentration. We consider that preferential formation of Sn-vacancy pairs is a key to reduce the concentration of electrically active vacancies. In addition, we found that the  $\text{H}_2$ -annealing reduces the Hall carrier concentration of Ge and  $\text{Ge}_{1-x}\text{Sn}_x$  thin layers.

We demonstrate the electrical properties at the interface between  $\text{Ge}_{1-x}\text{Sn}_x$  and a substrate. We prepared a pn junction with a  $\text{p}^+$ -doped  $\text{Ge}_{1-x}\text{Sn}_x$  epitaxial layer on n-Ge(001) substrate. Figure 2 shows the current density-voltage characteristics of the  $\text{p}^+$ - $\text{Ge}_{1-x}\text{Sn}_x$ /n-Ge junction for various measurement temperatures [5]. We found that the Sn incorporation into Ge does not generate any interfacial defects causing excess leakage current. Instead, the Sn incorporation reduces the leakage current at low temperature below 170K in the reverse bias condition compared to a  $\text{p}^+$ -Ge/n-Ge junction sample (*not shown*).

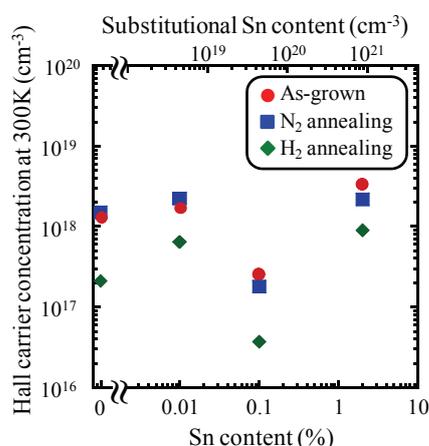
We also examined the epitaxial growth of  $\text{Ge}_{1-x-y}\text{Si}_x\text{Sn}_y$  ternary alloys on a Ge substrate [6]. We can deposit a  $\text{Ge}_{1-x-y}\text{Si}_x\text{Sn}_y$  epitaxial layer whose lattice constant was controlled to match bulk-Ge. Figure 3 shows the x-ray diffraction two dimensional reciprocal space map (XRD-2DRSM) for a  $\text{Ge}_{1-x-y}\text{Si}_x\text{Sn}_y$  layer grown on a Ge(001) substrate. The Si and Sn compositions were determined to be 58% and 15%, respectively, from Raman scattering spectroscopy and Rutherford back scattering methods. We achieved the pseudomorphic growth of  $\text{Ge}_{1-x-y}\text{Si}_x\text{Sn}_y$  ternary layer even with a high Sn content, since the strain of this layer to Ge is suppressed as small as 0.5% with local strain compensation between Si and Sn atoms. The full width at half maximum values of the  $\omega$  rocking curves of pseudomorphic  $\text{Ge}_{1-x-y}\text{Si}_x\text{Sn}_y$  layers are as small as that of a Ge substrate, indicating the superior crystalline structures of  $\text{Ge}_{1-x-y}\text{Si}_x\text{Sn}_y$  layers. Figure 4 shows the Ge-Si-Sn content dependence of the energy band gap calculated assuming Vegard's law. Ge-Si-Sn ternary alloy realizes the energy bandgap engineering with lattice-matching system on Ge [7]. We expected the energy bandgap of  $\text{Ge}_{1-x-y}\text{Si}_x\text{Sn}_y$  controlled to be from 0.66 eV of Ge to as high as 1 eV in this study.

### Acknowledgements

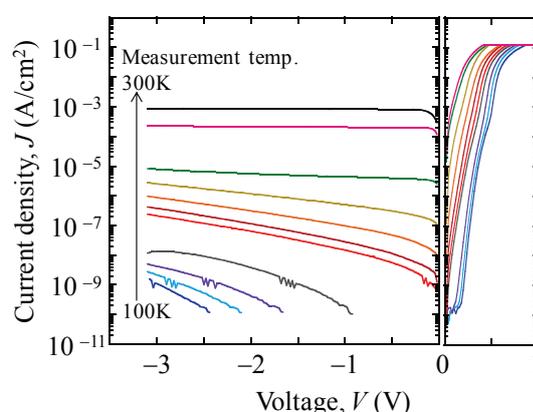
This work was partly supported by a Grant-in-Aid for Specially Promoted Research (No. 22000011) from the MEXT and the ALCA Program from JST in Japan.

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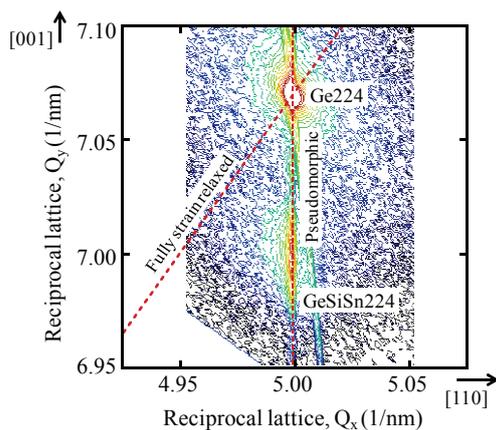
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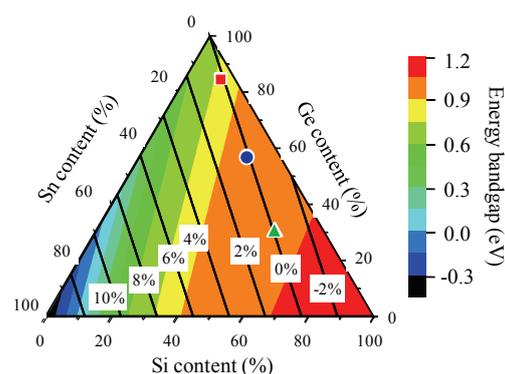
**Fig. 1** The Sn content dependence of the Hall concentration in undoped  $\text{Ge}_{1-x}\text{Sn}_x$  thin layers grown on SOI substrates for as-grown,  $\text{N}_2$ -annealed, and  $\text{H}_2$ -annealed samples [4].



**Fig. 2** The current density-voltage characteristics of the  $\text{p}^+\text{-Ge}_{1-x}\text{Sn}_x/\text{n-Ge}$  junction for various measurement temperatures.



**Fig. 3** XRD-2DRSM around the Ge224 Bragg reflection for a  $\text{Ge}_{1-x}\text{Si}_x\text{Sn}_y$  epitaxial layer grown on a Ge(001) substrate.



**Fig. 4** The Ge-Si-Sn content dependence of the calculated energy bandgap. The solid lines with percentages correspond to the lattice mismatch for bulk-Ge. Some symbols indicate the content conditions of samples prepared in our study.

## Epitaxial Growth of $\text{Ge}_{1-x}\text{Sn}_x$ with high Sn content by Reduced Pressure CVD

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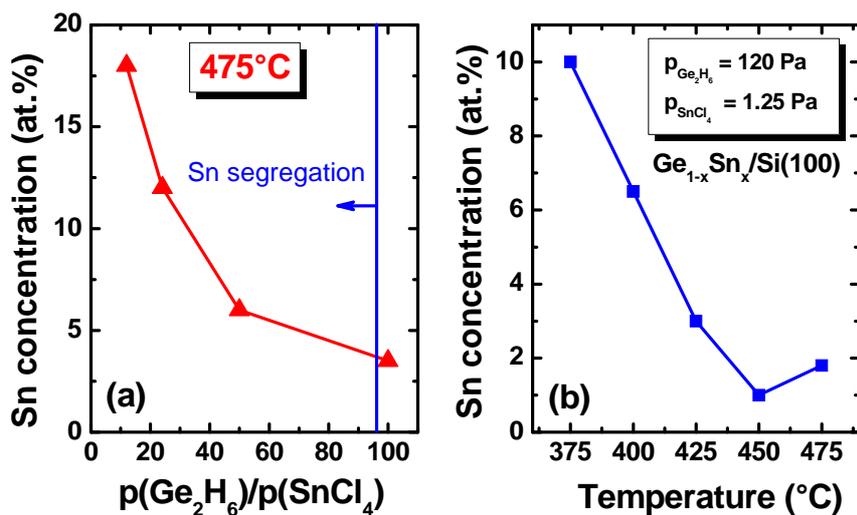
The most promising route towards photonic integrated circuits on Si relies on the development of appropriate group IV semiconductors, such as (Si)GeSn alloys. At high Sn contents these alloys present a direct energy band-gap [1] which is the requirement for efficient light emitters and detectors. Moreover, (Si)GeSn layers may act as buffer layers to introduce tensile strain in Ge [2] for high mobility channels of high performance MOSFETs. However, synthesizing these alloys is very challenging, because of the low solid solubility of Sn in Ge of about 1 at.%.

We are investigating epitaxial  $\text{Ge}_{1-x}\text{Sn}_x$  layers grown by low temperature Reduced Pressure CVD using an AIXTRON Tricent® cold wall system with a showerhead technology [3].  $\text{Ge}_2\text{H}_6$  and  $\text{SnCl}_4$  are employed as Ge and Sn precursors and  $\text{N}_2$  as carrier gas. At a reactor pressure of 60 mbar the  $\text{Ge}_{1-x}\text{Sn}_x$  layers were grown at susceptor temperatures between 375 - 475°C. The layers were studied by Rutherford Backscattering/Channeling (RBS/C) and Transmission Electron Microscopy (TEM) to analyze the layer stoichiometry, thicknesses and crystal quality. XRD measurements and Raman spectroscopy are employed to measure the strain status of the structures.

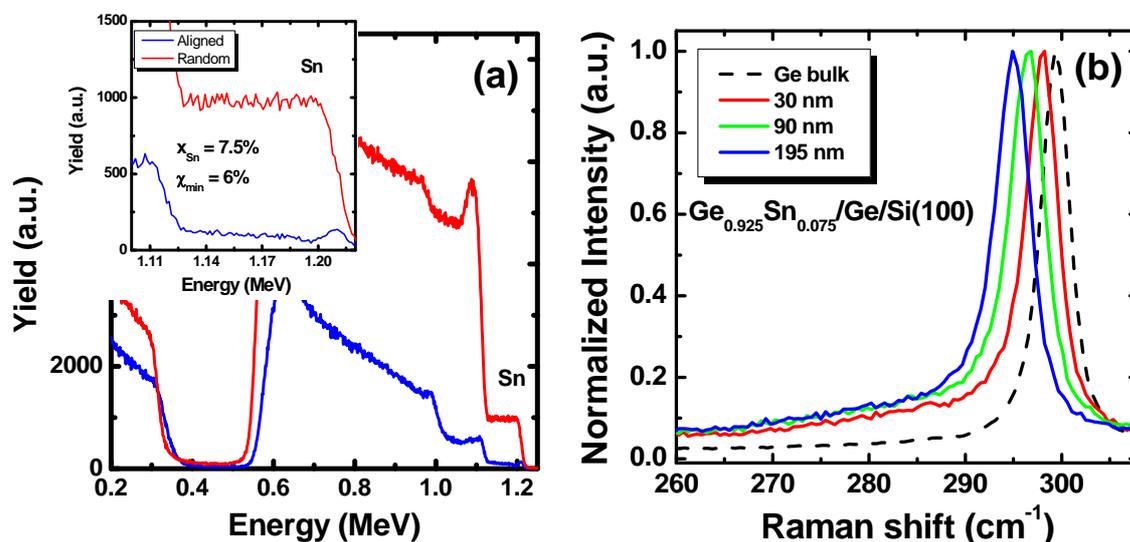
$\text{Ge}_{1-x}\text{Sn}_x$  layers were grown on Si(100) substrates and Ge buffer layers at different growth temperatures and Sn precursor fluxes in order to investigate GeSn growth properties on different substrates. In Fig. 1(a) the Sn concentration, measured by RBS, is shown as a function of the partial pressure ratio of  $\text{Ge}_2\text{H}_6$  and  $\text{SnCl}_4$  for layers grown at 475°C on Si(100). It is found, that decreasing the partial pressure ratio the Sn concentration increases from 1.8 % to about 18 %. However, SEM and RBS analysis reveal Sn segregation at these growth conditions for  $x > 1.8$  %. Using this optimized partial pressure ratio and decreasing the growth temperature to 375°C single crystalline layers with Sn concentration up to 10 % were grown (Fig.1b). Minimum yield values  $\chi_{\min}$  of 20 % prove a substitutional fraction of Sn atoms of about 80 %. The GeSn growth is significantly improved in terms of crystal quality and surface roughness using Ge buffered Si(100) substrates. Figure 2(a) shows RBS random and aligned spectra of a partially relaxed 195 nm thick  $\text{Ge}_{0.925}\text{Sn}_{0.075}$  layer grown on a Ge buffer layer. Both, high crystal quality and a high substitutionality of Sn  $> 90$  % are proved by a  $\chi_{\min}$  of 6 %. Raman spectra for three  $\text{Ge}_{0.925}\text{Sn}_{0.075}$  layers with thicknesses of 30 nm, 90 nm and 195 nm are presented in Fig. 2b. As a reference for the Ge-Ge modes a spectrum of Ge bulk sample is shown in black. The Raman shift towards lower wavenumbers for increasing layer thickness is due to the strain relaxation. This is confirmed by the TEM image in Fig. 3. For the partially relaxed  $\text{Ge}_{0.925}\text{Sn}_{0.075}$  layer dislocations are visible at the GeSn/Ge interface. FWHM values of about  $4 \text{ cm}^{-1}$  indicate high crystal quality. Moreover, the TEM image of the 30 nm thick  $\text{Ge}_{0.925}\text{Sn}_{0.075}$  layer (Fig. 3) shows no dislocations at the GeSn/Ge interface, indicating pseudomorphic GeSn growth on Ge(100).

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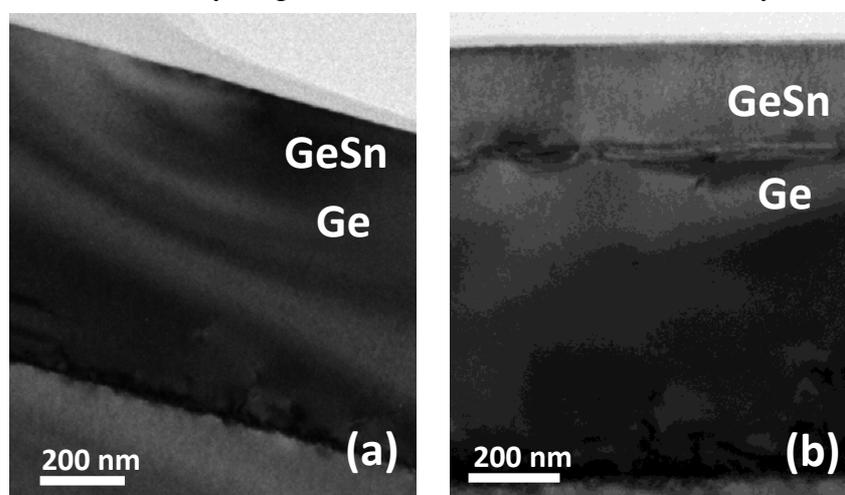
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**Fig. 1** (a) Sn concentration as a function of the partial pressure ratio of  $\text{Ge}_2\text{H}_6$  and  $\text{SnCl}_4$  grown at  $475^\circ\text{C}$  on  $\text{Si}(100)$ . For smaller ratios, strong Sn segregation is observed. (b) Sn concentration as a function of growth temperature at a fixed partial pressure ratio.



**Fig. 2** (a) Rutherford Backscattering (RBS) random (red) and aligned spectra and (b) Raman modes for three  $\text{Ge}_{0.925}\text{Sn}_{0.075}$  layers grown on a 600 nm thick Ge buffer layer.



**Fig. 3** TEM images of a (a) fully strained 30 nm thick  $\text{Ge}_{0.925}\text{Sn}_{0.075}$  layer and a (b) relaxed 195 nm thick  $\text{Ge}_{0.925}\text{Sn}_{0.075}$  layer grown on a Ge buffer layer.

## FIB preparation and TEM characterisation of Si/Sn alloys

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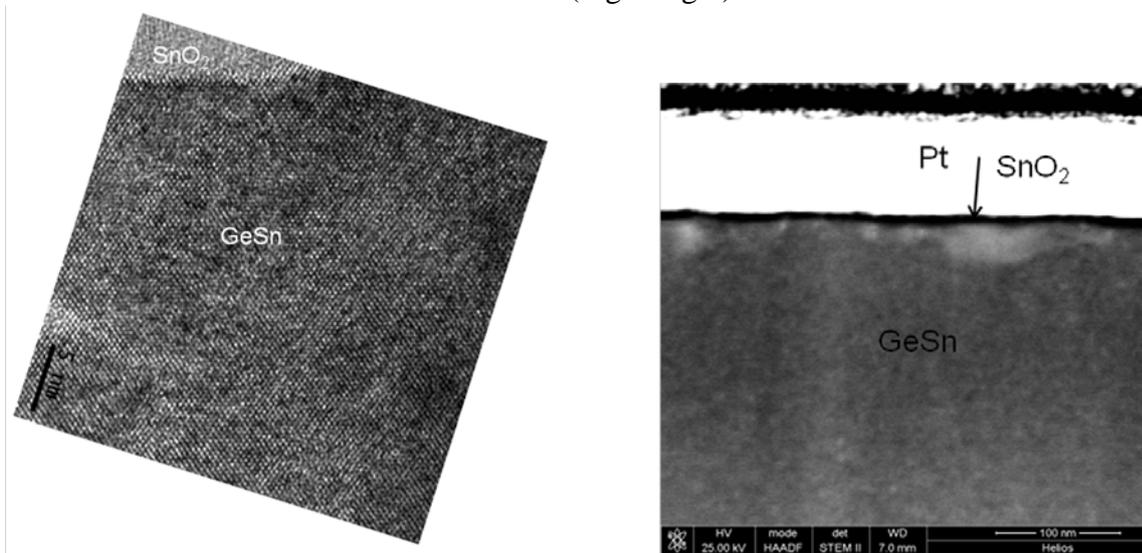
GeSn and SiGeSn alloys represents one of the most attractive group IV semiconductor materials, due to the higher carrier mobility they can achieve<sup>1</sup>.

It is therefore of paramount importance to be able to characterise them in terms of morphology, composition, cristallinity and strain distribution with a spatial resolution within the nanometre range. This can be nowadays routinely performed by Trasmision Electron Microscopy (TEM), but the specimen requirements for a successful TEM analysis are stringent: the sample has to be electron transparent, as little amorphised as possible and free from local thickness variations.

Focused Ion Beam (FIB) is a well known technique based on the use of a Ga<sup>+</sup> beam, capable of preparing thin TEM-ready lamellas from the region of interest, with a nanometre scale precision. However, in order to minimise the amorphisation of both the sides and the top of the thin lamella the ion currents have to be carefully chosen, depending on the electrical and mechanical characteristics of the analysed material.

GeSn alloys in particular are very sensitive to the ion beam, due to the instability of the cubic structure of  $\alpha$ -Sn and the low thermodynamic solubility of Sn in Ge (<1%)<sup>2</sup>

In this work, we briefly summarise the theoretical and practical aspects of the tecnique, discussing the main problems encountered and any possible solution. Various experimental results from GeSn specimens are also presented, proving that the FIB technique can not only provide suitable samples for High Resolution TEM analysis (Fig. 1 left) but also yield valuable information on the Sn distribution (Fig. 1 right).



**Fig. 1** HREM image of a GeSn sample, showing good crystallinity and little amorphisation (left); FIB-STEM image of a similar sample, with Sn clusters clearly visible near the GeSn/SnO<sub>2</sub> interface (right).

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## Laser processing of heteroepitaxial GeSn and SiGeSn alloys

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GeSn and SiGeSn alloys are attractive group IV semiconductor materials with superior carrier mobility, tunable band gaps in the midinfrared with indirect to direct transition above certain composition ranges<sup>1-3</sup>. The interesting properties of those alloys for independent adjustment of lattice parameters and band gap over a wide range stimulated the research dedicated to CMOS compatible novel device concepts including the combination of Si based with III/V based technologies<sup>4-6</sup>. Considerable efforts have been recently devoted on growing Ge-rich GeSn and SiGeSn alloys on Si, Ge and InP via MBE and CVD techniques<sup>7-10</sup>. The major problem for incorporating Sn into the Ge is the large lattice mismatch of 4.2% and 19.6% between Si ( $a_0=0.5431$  nm) and Ge ( $a_0=0.5658$  nm) or  $\alpha$ -Sn ( $a_0=0.6493$  nm), respectively, the instability of the diamond-cubic structure of  $\alpha$ -Sn and the low thermodynamic solubility of Sn in Ge (<1%)<sup>11</sup>.

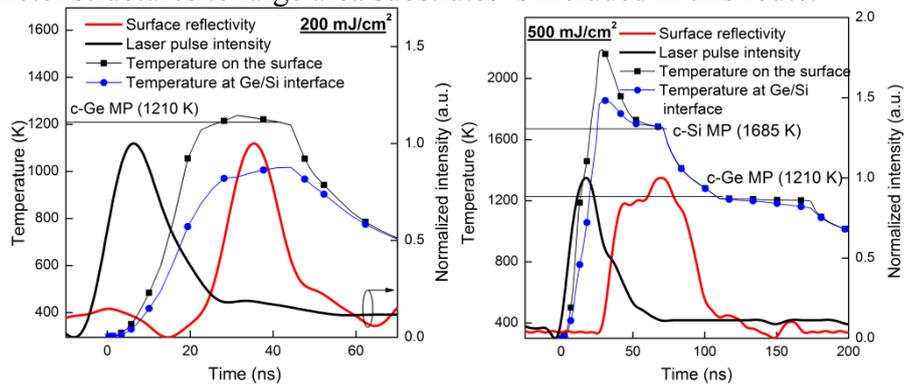
Because of these constraints conventional CVD and MBE growth is done at out of equilibrium conditions and low processing temperature that lead to Sn segregation<sup>7</sup>. To counteract these effects and improve the intermixing of the elements in order to obtain binary and ternary Si/Ge/Sn alloys with acceptable crystal quality we propose an alternative, out of equilibrium growth, based on ultra fast (several ns) laser melting/intermixing/epitaxial recrystallization cycles. This technique features low optical absorption depth and well controlled temperature gradient for ultra-rapid heating/cooling cycles, induced by pulses of about 25 ns, which provoke controlled changes in composition and structure of the material. These changes are directly related to the laser processing parameters energy density and number of pulses, thus determine the properties of the final product. An effective control of the experimental parameters is achieved simulating numerically the 1-D spatial thermal depth profile using finite element methods<sup>12-15</sup>. Laser processing of Ge/Sn layers was one of the first routes to metastable GeSn<sup>16</sup> with a strong lack of epitaxy. Pulsed Laser Induced Epitaxy (PLIE) is a technique that improves the epitaxial growth and has already proven to be effective for GeSn and SiGeSn alloys with Sn content below the solubility limits<sup>14,15,17</sup>.

Virtual Germanium (v-Ge) with different thickness on top of Si has been used for the GeSn and SiGeSn alloys. Sn concentration is controlled varying both processing parameters and the thickness of the top Sn layer deposited on the Ge. This layer was realized either after the v-Ge MBE deposition from Sn effusion cell. GeSn is obtained by intermixing of the Ge and Sn layers, whereas SiGeSn alloying results after intermixing of the Ge and Sn layers with the Si substrate. Samples were irradiated in Ar atmosphere, using 193 nm Excimer Laser (Lambda Physik LPX 220i). The laser beam was spatially homogenized using a fly-eye system (EX-HS-700D) and its energy was "in situ" controlled (Ophir PE50-DIF). Time Resolved Reflectivity (TRR) at 633 nm was used to monitor the phase change duration on the sample surface in real time. Experimental variables were energy density and number of laser shots. Analysis of the samples was performed by Raman spectroscopy (HJY LabRam HR800), TOF-SIMS (IonTOF TOF-SIMS-IV) and RBS.

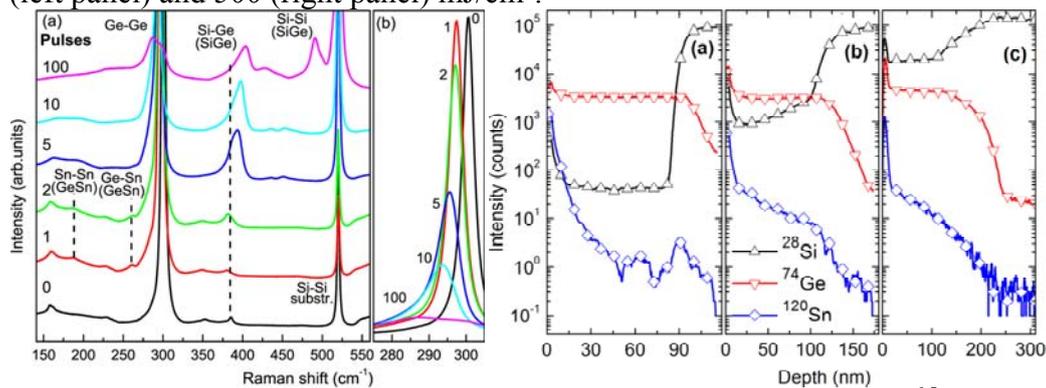
We started with a comparison of the melt duration extracted from the TRR signal and the one derived from the numerical solution of the heat conduction differential equation. The sample structure used for the numerical simulation was 4 nm Sn on top of 100 nm v-Ge on Si substrate. Melt depth and laser energy density, respectively, were adjusted depending if only the Sn and Ge layers melt (below 104 nm) is wished and GeSn is formed, or if also the Si substrate is included so that the resulting layer forms SiGeSn alloy. The estimated values were 200 mJ/cm<sup>2</sup> for GeSn and 500 mJ/cm<sup>2</sup> for SiGeSn (Fig. 1). However, the expected alloys were not obtained with one laser shot of these energy densities. We saw that the level of

intermixing of the corresponding layers (alloy components resp.) was strongly dependent on the number of laser shots (Fig. 2) and therefore samples irradiated with 100 pulses were further studied. Sn substitutional content from RBS yields 1% in GeSn and 0.6% in SiGeSn. We suggest that higher Sn content might be obtained varying Sn and Ge layer thickness that would reflect on the amount of incorporated Sn in both Ge and SiGe.

In summary, we report on an alternative technique to synthesize epitaxial GeSn and SiGeSn alloys and point out a route to reach Sn content above the solubility limit. Also, an improvement of the scalability of the applied technique for GeSn and SiGeSn alloys ranging from micrometer structures to large area substrates is included in this route.



**Fig. 1** Comparison of calculated and experimentally obtained melt duration for the first pulse of 200 (left panel) and 500 (right panel)  $\text{mJ}/\text{cm}^2$ .



**Fig. 2** Evolution of the level of intermixing with increasing number of pulses<sup>15</sup>. The left panel (a) shows the transition of the Raman modes from GeSn to SiGeSn for 0 to 100 pulses of 500  $\text{mJ}/\text{cm}^2$ , (b) represents the intensity of the Ge-Ge Raman mode that decreases with increasing intermixing of the three components. The right panel shows TOF-SIMS depth profile of the as received (a) irradiated with 5 shots (b) and 100 shots (c) of 500  $\text{mJ}/\text{cm}^2$  samples.

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## Fabrication of Titanium Oxide Nanotube Micro Gas Sensors by Anodization

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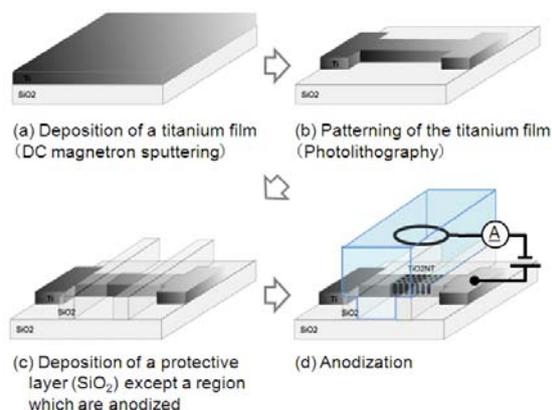
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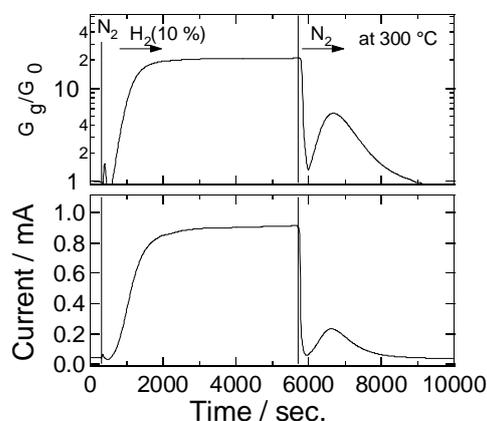
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The metal oxide semiconductor gas sensor is one of gas sensors and it has useful features of simple structure and high sensitivity. Miniaturization or integration of gas sensors gives a lot of advantages such as low power consumption, improvement of portability, high reliability, and simultaneous measurement of multicomponent. Miniaturization and integration of gas sensors require position controllability and uniformity of materials. It is important to precisely control the position of a device. Therefore, it is important to develop a hybrid process between photolithography and bottom-up process. This requires that the bottom-up process is compatible with photolithography. Then, we used an anodization process, which is compatible with photolithography, as a bottom-up one. The anodization process forms nanotubes with homogenous pore diameter and period. It has been reported that gas sensors using anodic titanium oxide nanotubes have good performance. In this study, we miniaturized metal oxide semiconductor gas sensors using an anodic titanium oxide nanotube film by the hybrid process.

Figure 1 illustrates the miniaturized gas sensor fabrication process. First, a titanium film was deposited on the substrate by DC magnetron sputtering and the titanium film was patterned to form the titanium wire. A protective silicon dioxide layer was deposited on the titanium wire except a region which was anodized. After that the part of the titanium wire was anodized in ammonium fluoride and water containing ethylene glycol solution. Finally, anodic titanium oxide nanotube layer was crystallized. Figure 2 shows the response of the miniaturized hydrogen gas sensor at 300 °C. From Fig. 2, we can see that the conductance change as large as the 20 times when the hydrogen concentration was 10 %. This indicates that a micro gas sensor can be fabricated by the anodization process. Furthermore, the sensed current of about 1 A indicated that a picoamperemeter and formation of interdigitated electrodes are not necessary for sensing current.



**Fig. 1** Fabrication process of micro gas sensors by local anodization of titanium wires



**Fig. 2** The response characteristics of a miniaturized hydrogen gas sensor at 300 °C.

## Application of titanium oxide nanotube films to solar cells

Michio Niwano, Ryota Kojima, Ma Teng, and Yasuo Kimura

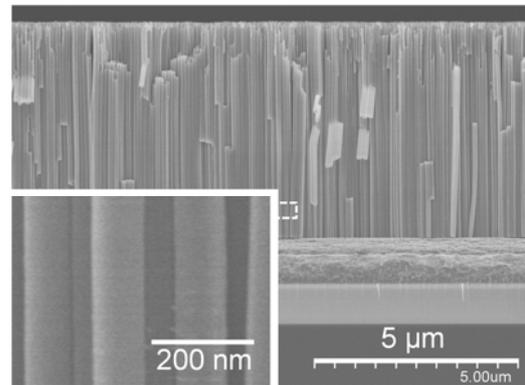
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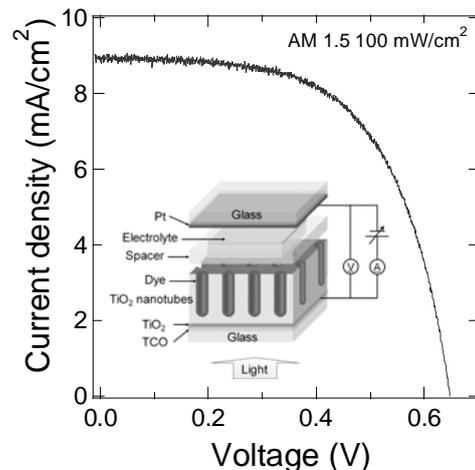
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Titanium oxide ( $\text{TiO}_2$ ) nanotubes attracts great interest because of their superior properties such as high photocatalytic activity and high biological affinity, and their application to solar cells, gas sensors and orthopedic implantation have been reported. Anodization of titanium (Ti) is a promising method to fabricate a  $\text{TiO}_2$  nanotube film on a substrate because it enables us to form a vertically oriented nanotube structure with uniform diameter and period. On the other hand, dye-sensitized solar cells (DSCs) have been widely studied because of their simple structure and relatively high conversion efficiency. DSCs are composed of a dye-adsorbed  $\text{TiO}_2$  nanoparticle film (negative electrode), electrolyte, and a counter electrode and a vertically oriented  $\text{TiO}_2$  nanotube film is an alternative to a  $\text{TiO}_2$  nanoparticle film because it satisfies the requirements for high energy conversion efficiency: straight current paths and a large surface area to adsorb a lot of dye molecules.

In this study, we fabricated an anodic  $\text{TiO}_2$  nanotube film on a transparent conductive oxide (TCO) layer by anodization of a Ti film in an ethylene glycol solution of water and ammonium fluoride ( $\text{NH}_4\text{F}$ ). The Ti film was deposited through the DC magnetron sputtering method. The anodization condition was optimized by adjusting electrolyte composition. Figure 1 shows a cross-sectional FE-SEM image of anodic  $\text{TiO}_2$  nanotube films formed by anodization of a Ti film at an anodic potential of 40 V in an ethylene glycol solution of 0.2 M  $\text{NH}_4\text{F}$  and 3.0 wt.% water. From Fig.1, we can see that a vertically oriented  $\text{TiO}_2$  nanotube film formed on a substrate. The tube diameter was about 90 nm. In addition, a  $\text{TiO}_2$  nanotube film was applied to a negative electrode of a DSC as shown in the inset of Fig. 2. The electrolyte of the DSC was a solution of 0.1 M LiI 0.05 M I<sub>2</sub>, 0.6 M DMPII, and 0.5 M TBP in acetonitrile. The counter electrode was a platinum thin film sputtered on a TCO layer. Figure 2 shows the  $I$ - $V$  characteristic of a DSC using an anodic  $\text{TiO}_2$  nanotube film as a negative electrode. This indicates that an anodic  $\text{TiO}_2$  nanotube film can be used as a negative electrode of DSC.



**Fig 1** A cross-sectional FE-SEM image of anodic titanium oxide nanotube films formed by anodization of Ti.



**Fig 2** The  $I$ - $V$  characteristic of a DSC using an anodic  $\text{TiO}_2$  nanotube film as a negative electrode.

## Dynamic Characteristics of Neuron Models and Microchip Integration of Active Neural Networks

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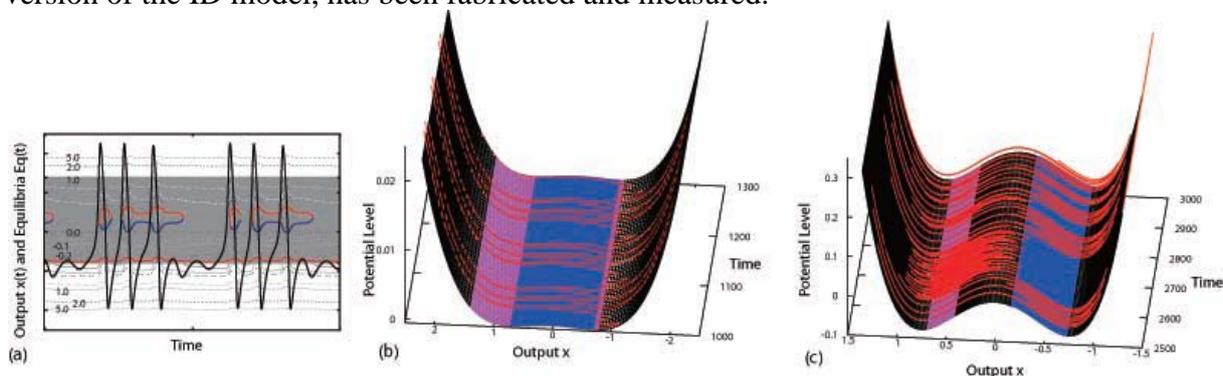
There have been many researches of various neuron models which typically take the form of ordinary nonlinear differential equations of several dimensions. The pattern of spiking is of great importance, because it is believed that it codifies the information transmitted by neurons. It is an actively studied problem to apply various neuron models to artificial neural networks (ANNs) for intelligent information processing in the field of nonlinear dynamics and the brain research.

One of important aspects of this situation is the lack of universal discussion over the dynamical behaviors of various neuron models, though perturbation and bifurcation theories exist. Thus we reveal that each model has its own potential function and active areas on the potential. Negative resistance is one of active areas. This concept realizes the universal discussion of the dynamical behaviors of models, for example, bursting, spiking, etc.

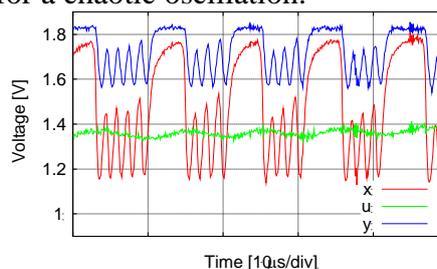
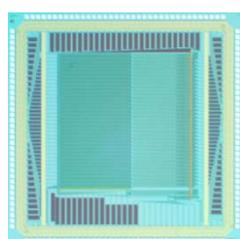
On the other hand, Hopfield neural network is capable of solving combinatorial optimization problems and it is a parallel-processing version of the gradient method. However, it has some drawbacks. One of the most concerning drawbacks is that it frequently finds locally minimum solutions instead of global minima.

The active areas of neuron models make the state of network escape from local minima by their destabilization. In computer simulation and theoretical estimation, we have already shown that the ID network, which is one of active ANNs to implement associative memory and combinatorial optimization problems, is capable of converging on optimal solutions only.

The prototype chip of a burst ID model, which shows bursting dynamics as an extended version of the ID model, has been fabricated and measured.



**Fig. 1** Burst and chaotic oscillations bound by the potential with the active areas. (a) Time series of output on a contour map of the potential. Bird's eye views of outputs on the potentials, (b) for a burst oscillation and (c) for a chaotic oscillation.



**Fig. 2** Photograph of a microchip for the Burst-ID ANN, and an observed bursting output.

## Ionic Liquid-Gated Organic Field-Effect Transistors

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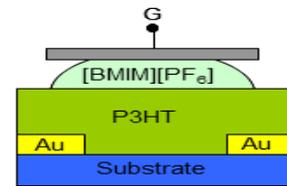
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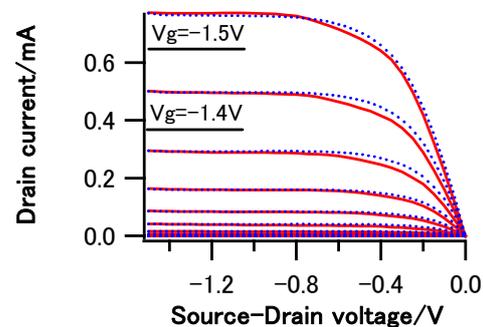
An ionic liquid-gated organic transistor can drive high output current at a low voltage compared with a general organic field effect transistor (OFET). It is necessary to elucidate the operation mechanism for improvement of the performance of ionic liquid-gated organic transistors. Two operation mechanisms of ionic liquid-gated organic transistors have been proposed. One is carrier generation due to electrochemical doping. The other is accumulation of a great number of carriers due to formation of an electric double layer, which functions as a very thin gate insulator. However, it is difficult to distinguish these mechanisms. In this study, the generation of carriers in a P3HT layer was in-situ observed using infrared absorption spectroscopy in the multiple internal reflection geometry (MIR-IRAS).

Figure 1 illustrates the structure of an ionic liquid-gated organic transistor. A gate insulator layer was replaced with ionic liquid. P3HT and [BMIM][PF<sub>6</sub>] were used as an organic layer and ionic liquid, respectively. The thickness of a P3HT layer was 15 nm. Figure 2 shows the typical output characteristic of ionic liquid-gated organic transistors. The gate length and width were 2  $\mu\text{m}$  and 1 mm, respectively. From Fig. 2, it can be seen that a large drain current of 0.7 mA flew when a gate voltage was 1.5 V and an ionic liquid-gated organic transistor can drive large current. Figure 3 shows infrared spectra or (a) a P3HT layer of the ionic liquid-gated organic transistor during operation and (b) a FeCl<sub>3</sub>-doped P3HT film [1]. The P3HT layer was 3  $\mu\text{m}$  in thickness. It was thick enough to observe penetration of PF<sub>6</sub><sup>-</sup> to the P3HT layer because it was much thicker than the penetration depth of the evanescent field of infrared. From Fig. 3, we can see that intensities of infrared absorption peaks due to P3HT cation increased with an increase in the gate voltage. This indicates that PF<sub>6</sub><sup>-</sup> penetrated into the P3HT layer to generate carriers, that is, electrochemical doping took place.

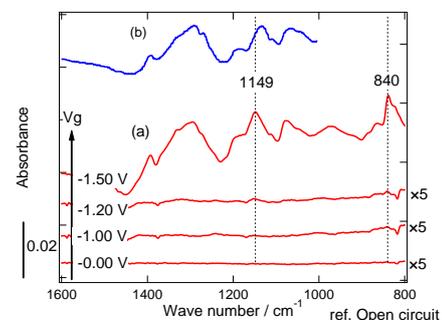
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**Fig 1** The structure of an ionic liquid-gated organic transistor.



**Fig 2** A typical output characteristic of ionic liquid-gated transistor.



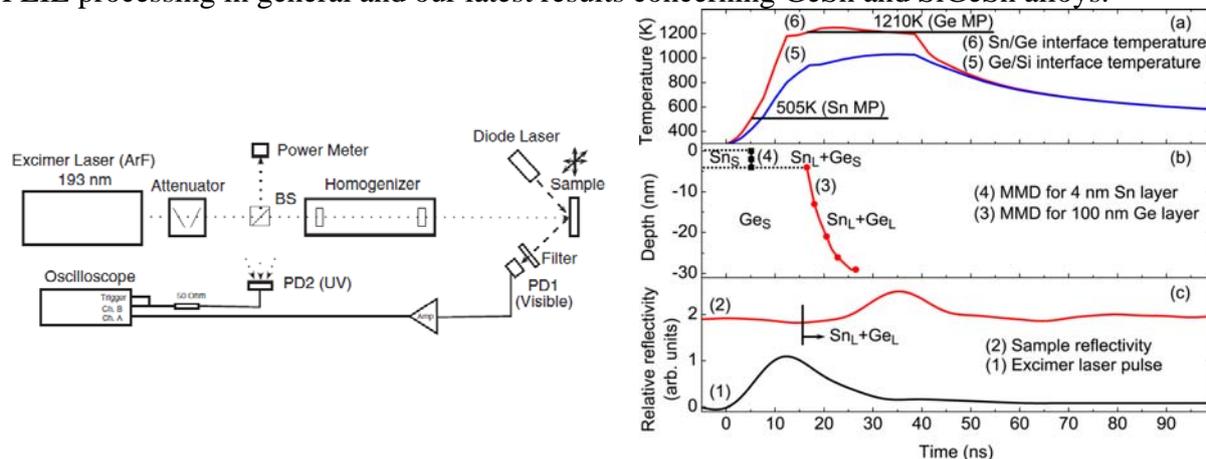
**Fig 3** Infrared spectra for (a) a P3HT layer of the ionic liquid-gated organic transistor during operation and (b) a FeCl<sub>3</sub>-doped P3HT film [1].

## UV Excimer laser processing of SiGe and SiGeSn: finite elements method numerical modeling

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During the years, Pulsed Laser Induced Epitaxy (PLIE) has always been the “alternative” for epitaxial formation of group IV alloys with great variety of intermixed components, ranging from SiGe<sup>1-3</sup>, SiC<sup>4</sup>, SiGeC<sup>5-7</sup> to GeSn<sup>8-10</sup> and SiGeSn<sup>9,11,12</sup>. First Abelson et al.<sup>1</sup> was able to overcome the drawback of PLIE, called “cellular” structure growth<sup>8,13</sup> for SiGe and our group for GeSn<sup>9</sup> (for the given bellow configuration), resulting epitaxial abrupt or tailored concentration Si/Ge/Sn layers. The process itself consists of ultra rapid (up to several hundred ns) pulsed laser heating/cooling of the alloy components previously deposited in form of a multilayer structure and a crystalline substrate followed by epitaxial growth and solidification of the mixture.

Numerical simulation of the laser/material interaction has been widely used, mainly in high power lasers processing, to predict and fine tune the optimal experimental conditions<sup>14,15</sup>. Same approach was later transferred to PLIE processing of group IV elements<sup>16</sup> and alloys<sup>10,12,17,18</sup>. In general, it consists of Finite Elements Method (FEM) solution of the Heat Conduction Differential Equation (HCDE) for given boundary conditions and known heat gradient devoted from the Beer–Lambert law. The FEM analysis yields the values of the maximum melting depth (MMD) and melts duration (MD) for each mesh element. Those parameters can be extracted directly compared with experimental ones after sample characterization. MD or liquid phase monitoring, the so called Time Resolved Reflectivity (TRR), of the ultra rapid PLIE process yields essential experimental information correlated to the level of intermixing and thus composition<sup>1,9,19</sup>. It can be easily obtained with the simple experimental setup shown on figure 1 (left). Fig. 1 (right) represents good agreement of the TRR signal with the data obtained from the FEM analysis of 4 nm Sn/100 nm Ge/Si(100) multilayer. This paper aims to represent our approach to FEM analysis of group IV elements PLIE processing in general and our latest results concerning GeSn and SiGeSn alloys.



**Fig. 1** Experimental PLIE/TRR setup (left). Temporal evolution of the (a) interface temperatures, (b) depth of solid-liquid transition, and (c) reflectivity, caused by the first laser pulse of 200 mJ/cm<sup>2</sup> (right).

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## Ion doped apatite coatings by laser ablation

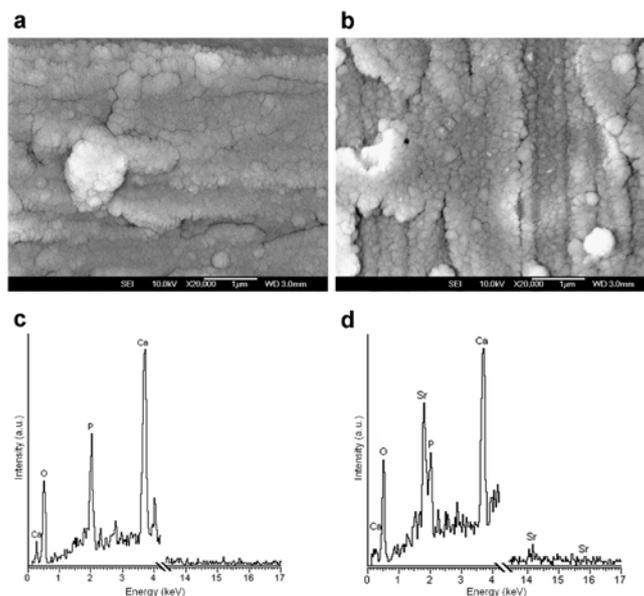
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The mineral family of apatites presents the property of being able to incorporate a variety of ions and compounds into its structure [2]. Hydroxyapatite,  $\text{Ca}_{10}(\text{PO}_4)_6(\text{OH})_2$  (HA), is a member of the apatites group which has been investigated extensively as a biomaterial due to its resemblance to the inorganic phase of bone. The growing evidence of the beneficial role of different ions (Si, Se,  $\text{Mg}^{2+}$ ,  $\text{K}^+$ ,  $\text{Na}^+$ ,  $\text{F}^-$ ,  $\text{Sr}^{2+}$ , ...) in bone tissue has increased the interest of developing ion doped apatites for medical applications, and specifically biocompatible coatings that can be deposited on metallic implants to benefit from their loadbearing capabilities, protection against the release of metal ions into the body and improvement of the implant fixation to hard tissues.

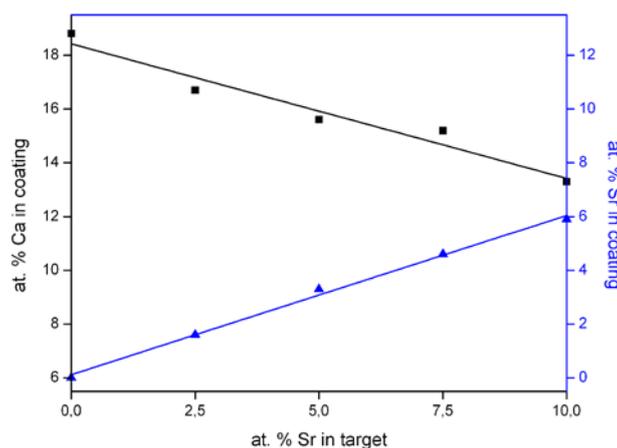
Pulsed laser deposition (PLD) is a technique that has proven effective in the fabrication of calcium phosphate films on metallic substrates with excellent final coating attachment. It makes use of laser pulses to expel target material from a rotating disk placed inside a vacuum reaction chamber and project it on the surface of a metallic substrate. The PLD method outperforms other techniques in versatility, since the choice of its ablation and deposition parameters control of the stoichiometry and crystallinity of the deposited material.

Ion doped hydroxyapatite (*i*HA:Si,Sr,Se) coatings have been fabricated by pulsed laser deposition (PLD) from initial targets obtained after mixing and compacting commercial HA and different proportions of precursors powders. The macroscopic morphology of the films presented in all cases equivalent spherical shaped aggregates of typical calcium phosphate coatings (see an example in Fig. 1). The incorporation of different ions (silicon, selenium, and strontium) into the HA compounds follows a linear behavior with the composition of the precursor compound in the original ablation target (Figs. 2 and 3).

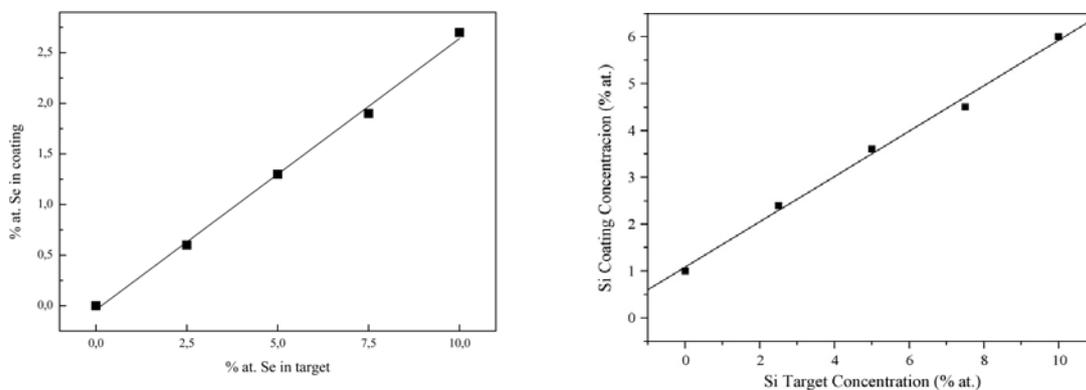
Additionally, the coatings thus fabricated were structurally, morphologically and chemically characterized using scanning electron microscopy, optical profilometry, X-ray diffraction, Fourier-transform infrared spectroscopy, X-ray photoelectron spectroscopy and energy dispersive X-ray spectroscopy. The role of the ion doped in the modification of the HA structure and possible substitutional mechanisms are discussed.



**Fig. 1** SEM micrographs and EDS analysis of two representative coatings deposited on titanium: pure HA (a,c) and *i*HA:Sr produced from an ablation target of 10% Sr (b,d).



**Fig. 2** Incorporation of Sr (▲) in the coating measured by XPS as a function of the at. % of Sr in the target, and consequent decremental effect of the Ca (■) content in the coatings.



**Fig. 3** Similar behaviors were found for *i*HA:Si and *i*HA:Se coatings obtained by laser ablation.

## Innovative porous SiC ceramics from natural precursors

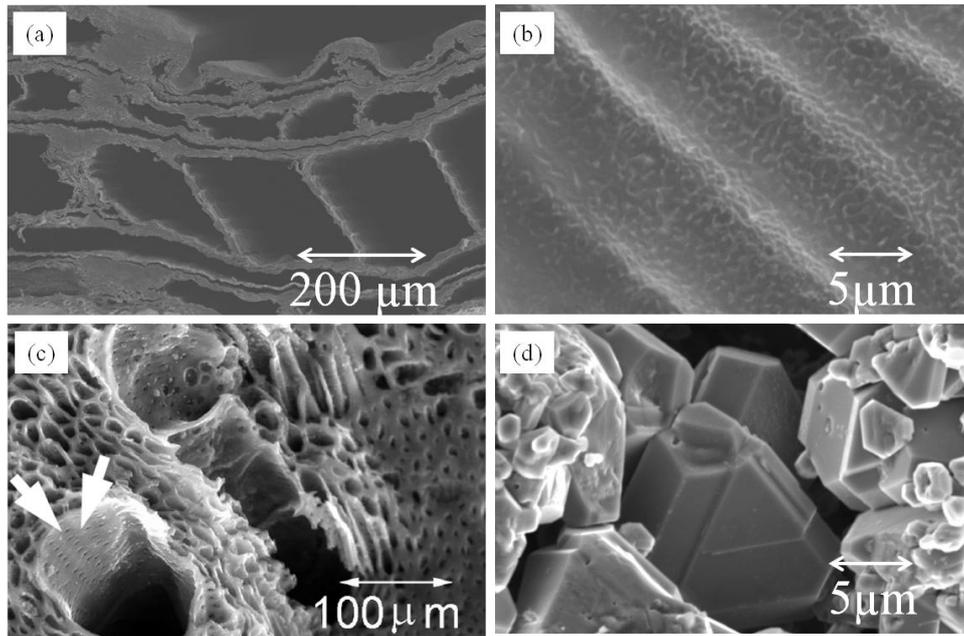
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Silicon carbide (SiC) belongs to the materials known as advanced ceramics which emerged as a consequence of the limitations of metal alloys for certain structural applications or in aggressive environments at high temperatures<sup>1,2</sup>. It is a low dense material ( $3.21\text{g/cm}^3$ ) composed by a covalent bond, covalence degree of 88%, and a similar structure to diamond. Both properties are the reasons for presenting a value of hardness of 13 in Mohr scale (being 15 for diamond<sup>1</sup>) as well as a great mechanical strength to both thermal shock and chemical attack<sup>3</sup>. At the same time, due to its electron configuration it acts as a semiconductor material. Silicon carbide can be obtained by different traditional methods as hot pressing sintering, pressureless sintering, chemical vapor deposition etc. The main drawbacks of the prior techniques are the complexity in the reproducibility of the pieces and the difficulty to obtain complex shapes, which reduces the range of applicability of the ceramics, the need for final machining and high temperatures requiring of expensive technology (high cost of manufacture) and the decrease of mechanical properties due to the use of additives<sup>1,3</sup>.

In 2001 a new method for the obtaining of SiC was developed<sup>4</sup>. This process met the requirements for all industrial manufacturing processes that are low cost and simplicity and, included the novel use of biological plant precursors as the carbon source<sup>4</sup>. It consists on the pyrolysis, or thermal decomposition, of the cellulose-based precursor to obtain a piece of porous carbon that preserves on it the interconnected vascular system of the plant. This piece of carbon is then subjected to a reactive infiltration with silicon (up to  $1550^\circ\text{C}$ ). The molten silicon will flow through the entire piece by means of the interconnected pores reacting with carbon, resulting in a porous SiC ceramic, also called Biomorphic<sup>®</sup> SiC<sup>1</sup>. The use of silicon powder of high purity or additives are not necessary and, due to the use of an open-porosity structure, the velocity of the synthesis increases together with the ability to produce pieces with complex shapes by modeling the carbon preform<sup>1</sup>.

Different cellulose-based precursors as trees (*Entandrophragma cylindricum*, *Eucalyptus* spp and *Pinus* spp), marine plants (*Juncus maritimus* and *Zostera marina*) or macroalgae species (*Laminaria ochroleuca*) have been successfully used for the obtaining of these porous SiC ceramics. The fact that the final ceramic retains the biological porosity and interconnectivity is one of the reasons for the great potential of this material, as traditionally it has been “printed” on the different materials the specific needed microstructure. With the use of these plant precursors, that costly process is avoided given that the desired microstructure has been already developed on the materials by nature<sup>6</sup>, providing the ceramics with macro- and micro-patterning on their surfaces, ranges of porosity from nano to macro-scale... that will make it possible to select one over another depending on the final application<sup>5</sup>. Thus, it has been introduced in industrial processes as a compound resistant to abrasion and corrosion (mechanical valves), to friction (brake pads), to temperature (receivers of solar radiation, engines, turbines), also in structural reinforcing concretes, acoustic and thermal isolation... or even as a biomaterial for bone prosthesis<sup>1,5,7</sup>. As a semiconductor material it is also attractive for blue light emitting diodes, ultraviolet and high-temperature sensors, radiation-hardened electronics, high-power electronics... This work pretends to summarize the great variety of materials with a range of microstructures and porosities that these biological precursors have provided to the ceramics field.



**Fig. 1** Scanning electron micrographs presenting porosity (a, c), surface patterning (b) of different cellulose-based precursors and a crystal of silicon carbide (d).

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